

**Section 2-3 Carbon Compounds (pages 44-48)****Key Concept**

- What are the functions of each group of organic compounds?

**The Chemistry of Carbon (page 44)**

1. How many valence electrons does each carbon atom have? \_\_\_\_\_  
\_\_\_\_\_
2. What gives carbon the ability to form chains that are almost unlimited in length?  
\_\_\_\_\_  
\_\_\_\_\_

**Macromolecules (page 45)**

3. Many of the molecules in living cells are so large that they are known as \_\_\_\_\_.
4. What is the process called by which macromolecules are formed? \_\_\_\_\_
5. When monomers join together, what do they form? \_\_\_\_\_
6. What are four groups of organic compounds found in living things?
  - a. \_\_\_\_\_
  - b. \_\_\_\_\_
  - c. \_\_\_\_\_
  - d. \_\_\_\_\_

**Carbohydrates (pages 45-46)**

7. What atoms make up carbohydrates? \_\_\_\_\_  
\_\_\_\_\_
8. Circle the letter of each sentence that is true about carbohydrates.
  - a. Starches and sugars are examples of carbohydrates.
  - b. Living things use them as their main source of energy.
  - c. The monomers in sugar polymers are starch molecules.
  - d. Plants and some animals use them for strength and rigidity.
9. Single sugar molecules are also called \_\_\_\_\_
10. Circle the letter of each monosaccharide.
  - a. galactose
  - b. glycogen
  - c. glucose
  - d. fructose

## Section 3 2 Review Carbon Compounds Answers

**E Durkheim**



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