

6. What type of bond(s) were formed in the compound from question #4?

Ionic Bond

7. Draw the bond(s) that will form between P and Cl.



8. What is the formula for the compound formed in question #7?



9. What type of bond(s) were formed in the compound from question #7?

Covalent Bond

10. Draw the correct valence electron configuration for each atom given.

<i>H</i>	<i>I</i>	<i>Ca</i>	<i>Al</i>
<i>Ne</i>	<i>Ge</i>	<i>B</i>	<i>K</i>
<i>Li</i>	<i>Si</i>	<i>C</i>	<i>Xe</i>

11. Fill in the table with the correct values.

Element	# valence electrons	# covalent bonds able to be formed
H	<i>1</i>	<i>1</i>
He	<i>2</i>	<i>0</i>
C	<i>4</i>	<i>4</i>
Cl	<i>7</i>	<i>1</i>
Ar	<i>8</i>	<i>0</i>
O	<i>6</i>	<i>2</i>
N	<i>5</i>	<i>3</i>
P	<i>5</i>	<i>3</i>

8 - Valence e⁻ Correct

Section 9 1 The Covalent Bond Answer Key

Lingjun Ying



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Table of Contents Section 9 1 The Covalent Bond Answer Key

1. Understanding the eBook Section 9 1 The Covalent Bond Answer Key
 - The Rise of Digital Reading Section 9 1 The Covalent Bond Answer Key
 - Advantages of eBooks Over Traditional Books
2. Identifying Section 9 1 The Covalent Bond Answer Key
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Section 9 1 The Covalent Bond Answer Key
 - User-Friendly Interface
4. Exploring eBook Recommendations from Section 9 1 The Covalent Bond Answer Key
 - Personalized Recommendations
 - Section 9 1 The Covalent Bond Answer Key User Reviews and Ratings
 - Section 9 1 The Covalent Bond Answer Key and Bestseller Lists
5. Accessing Section 9 1 The Covalent Bond Answer Key Free and Paid eBooks
 - Section 9 1 The Covalent Bond Answer Key Public Domain eBooks
 - Section 9 1 The Covalent Bond Answer Key eBook Subscription Services
 - Section 9 1 The Covalent Bond Answer Key Budget-Friendly Options
6. Navigating Section 9 1 The Covalent Bond Answer Key eBook Formats
 - ePub, PDF, MOBI, and More
 - Section 9 1 The Covalent Bond Answer Key Compatibility with Devices
 - Section 9 1 The Covalent Bond Answer Key Enhanced eBook Features
7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Section 9 1 The Covalent Bond Answer Key
 - Highlighting and Note-Taking Section 9 1 The Covalent Bond Answer Key
 - Interactive Elements Section 9 1 The Covalent Bond Answer Key

8. Staying Engaged with Section 9 1 The Covalent Bond Answer Key
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Section 9 1 The Covalent Bond Answer Key
9. Balancing eBooks and Physical Books Section 9 1 The Covalent Bond Answer Key
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Section 9 1 The Covalent Bond Answer Key
10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
11. Cultivating a Reading Routine Section 9 1 The Covalent Bond Answer Key
 - Setting Reading Goals Section 9 1 The Covalent Bond Answer Key
 - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Section 9 1 The Covalent Bond Answer Key
 - Fact-Checking eBook Content of Section 9 1 The Covalent Bond Answer Key
 - Distinguishing Credible Sources
13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
14. Embracing eBook Trends
 - Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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