

# Chapter 5: Chemical changes 1

## Knowledge organiser

### Reactions of metals

The reactivity of a metal is how chemically reactive it is. It is often related to water, as some metals react with it. Metals that react with water are highly reactive. Other metals will barely react with water or not at all, so they are less reactive.

Reaction with water	Reaction with acid	Reactivity series	Displacement reactions
Some group 1 hydrogen gas	Hydrogen	Most reactive	Hydrogen
Some group 2 hydrogen gas	Hydrogen	High reactivity	
Some group 2 hydrogen gas	Hydrogen	Medium reactivity	
Some group 2 hydrogen gas	Hydrogen	Low reactivity	
Some group 2 hydrogen gas	Hydrogen	Low reactivity	
Some group 2 hydrogen gas	Hydrogen	Low reactivity	
Some group 2 hydrogen gas	Hydrogen	Low reactivity	
Some group 2 hydrogen gas	Hydrogen	Low reactivity	
Some group 2 hydrogen gas	Hydrogen	Low reactivity	
Some group 2 hydrogen gas	Hydrogen	Low reactivity	

### Metal extraction

Some metals, like gold, are so unreactive that they are found as pure metals in the Earth's crust and can be mined.

Most metals will be compounds in the Earth's crust and have to be extracted from the rock. This is done by using a process called reduction.

Metals that are less reactive than carbon can be extracted by reduction with carbon. For example:

Metals that are more reactive than carbon can be extracted using a process called electrolysis.

### Salts

When acids react with metals or metal compounds, they form salts. It can be a compound where the hydrogen from the acid has been replaced by a metal. For example, when acid reacts with sodium hydroxide, the acid is replaced with the sodium salt.

Acid	Salt	Acid	Salt	Acid	Salt
Hydrochloric acid	HCl	Sulfuric acid	H <sub>2</sub> SO <sub>4</sub>	Nitric acid	HNO <sub>3</sub>
Hydrobromic acid	HBr	Hydrofluoric acid	HF	Carbonic acid	H <sub>2</sub> CO <sub>3</sub>
Hydroiodic acid	HI	Hydrocyanic acid	HCN	Acetic acid	CH <sub>3</sub> COOH
Hydroperchloric acid	HPClO <sub>4</sub>	Hydroperfluoric acid	HPF <sub>6</sub>	Hydroperbromic acid	HPBrO <sub>4</sub>

# Chapter 5: Chemical changes 2

## Knowledge organiser

### Reactions of acids

Acids react with some metals to form salts and hydrogen gas.

Acids react with some metals to form salts and hydrogen gas.

### Reactions of acids with metal hydroxides

Acids react with metal hydroxides to form salts and water.

Acids react with metal hydroxides to form salts and water.

### Reactions of acids with metal oxides

Acids react with metal oxides to form salts and water.

Acids react with metal oxides to form salts and water.

### Balancing symbol equations

When writing symbol equations you need to ensure that the number of each atom is the same on both sides.

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### Alkalis and bases

Alkalis are bases that dissolve in water to form hydroxide ions.

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### Strong and weak acids

Strong acids are acids that completely dissociate in water.

Strong acids are acids that completely dissociate in water.

### Concentrated and dilute acids

Concentrated acids have a high concentration of acid molecules.

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### State symbols

State symbols are used to indicate the physical state of a substance.

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### Displacement reactions

A displacement reaction is a reaction where one element replaces another in a compound.

### Ionic equations (HIT only)

Ionic equations are used to show the reaction between ions in a solution.

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### Half equations (HIT only)

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### Reactivity and ions

The reactivity of a metal is how chemically reactive it is. It is often related to water, as some metals react with it.

### Steps for writing an ionic equation (HIT only)

Steps for writing an ionic equation are as follows:

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### Acids and alkalis

Acids are compounds that, when dissolved in water, release H<sup>+</sup> ions.

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# Section Chemical Changes Answers

**Jicheng Xie**



## Section Chemical Changes Answers:

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content areas for which the present workbook is having some inputs Students of class six should have a proper understanding of basic shapes number system daily life problems and ecological concerns Most of the problems are related to daily experiences and normal operational concerns It is expected that students should go on facing day to day problems from science mathematics and humanities They should also address problems related to high order thinking skills They also participate in online digital classes and social media platforms for exploring relevant information on certain topic Hunting merely for information may not fulfill the purpose in particular Information duly collected should have adequate alignment with facts and figures for ensuring the process of remembering and recollecting such kinds of learning during need We are also incorporating few words from the faculty of mathematics Most of the part of publication is based on the pattern of questions people select for Olympiads Talent Search Examinations and other competitive examinations of similar nature This publication also introduces a learner with some apprehensions of Critical thinking Mathematics deals with some fundamental aspects related to time and space We all learn different rules and related operations starting from our elementary stage of schooling Different students take the subject differently as per their interest and willingness Some students calculate values with adequate speed and some other students do the same with lot of difficulties We also point out the development of fear related to Mathematics in the mind of some of the fellow students We cannot analyse the possible reasons of the development of such fear in the mind of students This development cannot be generalised It is not developed in the minds of all the fellow students Things often become difficult when our fellow ward fail to correlate the linkages of real life problems with that of mathematical ones It is the main reason of the lack of proper orientation in the process of the development of mathematical skills A skillful student can correlate both the aspects of mathematics and real life problems with much efficiency A skillful student of mathematics should be a good observer a perfect planner optimum analyzer and abled calculator Some students can take much time in solving any individual mathematical problem that compared to the time taken by the other fellow from the same peer group This book is designed to expose a student to different types of mathematical problems from the allied fields of the curriculum specified for the middle school It is expected that this workbook can equip a student in different ways and enable them to acquire mathematical skills with a long lasting impression in mind

Science for Ninth Class Part 1  
Chemistry Lakhmir Singh & Manjit Kaur, A series of books for Classes IX and X according to the CBSE syllabus and CCE Pattern

**A History of Chemical Theories and Laws** Matthew Moncrieff Pattison Muir, 1907      **Journal of the Society of Arts** Royal Society of Arts (Great Britain), 1864      **Science Examinations ... Reports, Etc** , 1904      *Common Core Science 4 Today, Grade 5* , 2014-05-15 Common Core Science 4 Today Daily Skill Practice provides the perfect standards based activities for each day of the week Reinforce science topics and the math and language arts Common Core State Standards all year long in only 10 minutes a day Weeks are separated by science topic so they may be completed in the order that best complements your science curriculum Review essential skills during a four day period and assess on the fifth day

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**Journal**, 1864 **Pharmaceutical Journal**, 1906 **Journal of the Royal Society of Arts** Royal Society of Arts (Great Britain), 1864 Journal Royal Society for the Encouragement of Arts, Manufactures and Commerce, 1864

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