

NETWORK SECURITY LAB MANUAL



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For VI Semester (CS &E) Diploma Engineering

GOVERNMENT POLYTECHNIC,

KARWAR

2014-2015

Network Security And Cryptography Lab Manual

Jonathan S. Weissman



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lover

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web 10 rue gustave flaubert 10400 nogent sur seine 33 0 3 25 24 76 34 contact museecamilleclaudel fr 1 hour from the gare de l est station in paris more admission full rate admission 8 per person reduced rate 4 50 per person free admission for visitors under 26 years of age

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550 j Δh q water 550 j

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calorimeter with a total heat capacity of 15.48 kJ/C the temperature of the calorimeter and its contents changes from 19.77°C to 33.90°C calculate q and h for the following reaction as written $2\text{NH}_3(\text{g}) + 3\text{O}_2(\text{g}) \rightarrow 2\text{H}_2\text{O}(\text{g}) + 3\text{H}_2(\text{g})$

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web choose 1 answer the final temperature is closer to t_1 than to t_2 a the final temperature is closer to t_1 than to t_2 the final temperature is exactly halfway between t_1 and t_2 b the final temperature is exactly halfway between t_1 and t_2 the final temperature is closer to t_2 than to t_1 c

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web 6 a 218 g sample of steam at 121°C is cooled to ice at 14°C find the change in heat content of the system 7 in going from ice at 34°C to steam at 138°C a sample of water absorbs 1.41×10^5 J find the mass of the sample answers 1 2 68 $\times 10^4$ J 2 6 23 $\times 10^4$ J 3 1 71 $\times 10^4$ J 4 1 11 $\times 10^5$ J 5 2 11 $\times 10^5$ J 6

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web there's 2 on the left and 1 on the right so we need to change the coefficient of NH_3 to 2 now we have $\text{N}_2 + \text{H}_2 \rightarrow 2\text{NH}_3$ total the atoms up again on the left there is 2 N and 2 H still on the right there is 2 N and 6 H now so now all we need to do is make the left side have 6 H in total

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