

91. Under the Indian Constitution, concentration of wealth violates

- (a) the Right to Equality ✓
- (b) the Directive Principles of State Policy
- (c) the Right to Freedom ✗
- (d) ~~the Concept of Welfare~~

92. What is the position of the Right to Property in India?

- (a) ~~Legal right available to citizens only~~
- (b) Legal right available to any person
- (c) Fundamental Right available to ~~all~~ citizens only
- (d) Neither Fundamental Right nor legal right

93. What was the exact constitutional status of India on 26th January, 1950?

- (a) A Democratic Republic
- (b) ~~A Sovereign Democratic Republic~~
- (c) A Sovereign Secular Democratic ✗ Republic
- (d) A Sovereign Socialist Secular ✗ Democratic Republic

94. Constitutional government means

- (a) a representative government of a nation with federal structure ✗
- (b) a government whose Head enjoys ✗ nominal powers
- (c) a government whose Head enjoys + real powers
- (d) ~~a government limited by the terms of the Constitution~~

95. With reference to India, the terms 'Halbi, Ho and Kui' pertain to

- (a) dance forms of Northwest India ✗
- (b) musical instruments
- (c) pre-historic cave paintings ✗
- (d) ~~tribal languages~~

96. Consider the following statements in respect of Bharat Ratna and Padma Awards :

1. Bharat Ratna and Padma Awards are titles under the Article 18(1) of the Constitution of India. ✓
2. Padma Awards, which were instituted in the year 1954, were ✓ suspended only once.
3. The number of Bharat Ratna Awards is restricted to a maximum ✓ of five in a particular year.

Which of the above statements are not correct?

- (a) 1 and 2 only ✓
- (b) ~~2 and 3 only~~
- (c) 1 and 3 only ✓
- (d) 1, 2 and 3 ✓

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web to prepare 1000 ml of a 0.1 mol/l solution of potassium nitrate we have to dissolve 10.1102 g of kno<sub>3</sub> 100 purity in deionized or distilled water after the solid is

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web sec 16.5 page 743 solution sec 12.3.12.4 page 519 in a saturated potassium nitrate kno<sub>3</sub> solution in water h<sub>2</sub>o a dynamic equilibrium will be established and the

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web jul 4 2023 density of saturated kno<sub>3</sub> solution what mass g of kno<sub>3</sub> will crystallize out of solution if exactly 100 g of its saturated solution 75°C is cooled to 25°C 2nd

**a solution contains 35 grams of socratic** - Aug 25 2022

web jan 1 2021 this study reports the density and viscosity measurement of ternary salts consisting of ca no<sub>3</sub> 2 nano<sub>3</sub> and kno<sub>3</sub> the addition of calcium nitrate has been

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web the amount of kno<sub>3</sub> in the solution is called solubility if the kno<sub>3</sub> is less than what the water can handle then it is an unsaturated solution if the amount of kno<sub>3</sub> is equal to

**100g of a saturated solution of kno<sub>3</sub> was cooled from 80** - Dec 17 2021

**potassium nitrate solution** - Sep 25 2022

web oct 24 2020 the solubility of kno<sub>3</sub> in 100g of water is 32 at 20°C and 109 at 60°C i have a doubt on 20g water is evaporated from 200g kno<sub>3</sub> statement i'm assuming that as

**finding mass percent through molality of potassium nitrate solution** - Jan 30 2023

web nov 4 2016 potassium nitrate is exceptionally soluble in water your solution is unsaturated a solution is saturated if it contains the same amount of solute as

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