

## Unit 1

(1.1)

### Structure, Bonding and Intro to organic chemistry.

#### 1.1 -- Formulae, equations and amount of Substance.

##### Atom --

It is that small particle of a substance which can take part in a chemical reaction, for example, if we have a small piece of iron metal, then every atom is capable of reacting with  $O_2$ , and rusting away. The atom of element hydrogen is represented by  $H$ , and that of Chlorine is represented by  $Cl$ . This means that an atom is the smallest unit of ordinary matter that forms a chemical element.

##### Element --

It is a pure substance that cannot be broken down into any further simpler substances. In our example of the small piece of iron metal, there are only iron atoms, and hence, it cannot be broken down into other simpler substances.

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### **Table of Contents Wch01 01 January 2015 Markscheme**

1. Understanding the eBook Wch01 01 January 2015 Markscheme
  - The Rise of Digital Reading Wch01 01 January 2015 Markscheme
  - Advantages of eBooks Over Traditional Books
2. Identifying Wch01 01 January 2015 Markscheme
  - Exploring Different Genres
  - Considering Fiction vs. Non-Fiction
  - Determining Your Reading Goals
3. Choosing the Right eBook Platform
  - Popular eBook Platforms
  - Features to Look for in an Wch01 01 January 2015 Markscheme
  - User-Friendly Interface
4. Exploring eBook Recommendations from Wch01 01 January 2015 Markscheme
  - Personalized Recommendations
  - Wch01 01 January 2015 Markscheme User Reviews and Ratings

- Wch01 01 January 2015 Markscheme and Bestseller Lists
- 5. Accessing Wch01 01 January 2015 Markscheme Free and Paid eBooks
  - Wch01 01 January 2015 Markscheme Public Domain eBooks
  - Wch01 01 January 2015 Markscheme eBook Subscription Services
  - Wch01 01 January 2015 Markscheme Budget-Friendly Options
- 6. Navigating Wch01 01 January 2015 Markscheme eBook Formats
  - ePub, PDF, MOBI, and More
  - Wch01 01 January 2015 Markscheme Compatibility with Devices
  - Wch01 01 January 2015 Markscheme Enhanced eBook Features
- 7. Enhancing Your Reading Experience
  - Adjustable Fonts and Text Sizes of Wch01 01 January 2015 Markscheme
  - Highlighting and Note-Taking Wch01 01 January 2015 Markscheme
  - Interactive Elements Wch01 01 January 2015 Markscheme
- 8. Staying Engaged with Wch01 01 January 2015 Markscheme
  - Joining Online Reading Communities
  - Participating in Virtual Book Clubs
  - Following Authors and Publishers Wch01 01 January 2015 Markscheme
- 9. Balancing eBooks and Physical Books Wch01 01 January 2015 Markscheme
  - Benefits of a Digital Library
  - Creating a Diverse Reading Collection Wch01 01 January 2015 Markscheme
- 10. Overcoming Reading Challenges
  - Dealing with Digital Eye Strain
  - Minimizing Distractions
  - Managing Screen Time
- 11. Cultivating a Reading Routine Wch01 01 January 2015 Markscheme
  - Setting Reading Goals Wch01 01 January 2015 Markscheme
  - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Wch01 01 January 2015 Markscheme
  - Fact-Checking eBook Content of Wch01 01 January 2015 Markscheme
  - Distinguishing Credible Sources

13. Promoting Lifelong Learning
  - Utilizing eBooks for Skill Development
  - Exploring Educational eBooks
14. Embracing eBook Trends
  - Integration of Multimedia Elements
  - Interactive and Gamified eBooks

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